S V R K GOVERNMENT DEGREE COLLEGE :: NIDADAVOLE TABLE – A – CURRICULAR PLAN – LECTURER WISE

NAME OF THE LECTURER

: Dr V. V. Ravindra

DEPARTMENT : CHEMISTRY

CLASS: III B.SC (B2C) YEAR: 2022-2023

SEMESTER: V PAPER : V11(B)

SERIAL NUMBER	MONTH & WEEK	HOURS AVAILABLE	SYLLABUS TOPIC	PUT	CURRICULAR ACTIVITY				CO-CURRICULAR ACTIVITIY				
				ADDITIONAL INPUT N'ALUE ADDITION	ACTIVITY	HOURS	WHETHER	IF NOT, ALTERNATIV E DATE	ACTIVITY	HOURS ALLOTED	WHETHER CONDUCTED	IF NOT, ALTERNATIV E DATE	REMARKS
1	2	3	4	5	6	7	8	9	10	11	12	13	14
	3rd week	4	Unit-1: Chromatography -Introduction and classification Principle, Classification of chromatographic methods	Differences between R _f values.	Lecture	4							
	4th week	4	. Nature of adsorbents, eluents, Rfvalues, factors affecting Rfvalues		Demonstration	3			Audio visual quiz	1			
	1st week	4	Unit-2: TLC and paper chromatography 1. Thin layer chromatography: Principle, Experimental procedure, preparation of plates, adsorbents and solvents,	Applications of TLC	Power point	4							
Dec	2nd week	4	Development of chromatogram, detection of spots, applications and advantages. 2. Paper Chromatography: Principle, Experimental		Digital class	3			Assign ment	1			
	3rd week	4	procedure, choice of paper and solvents, various modes of development- ascending, descending, radial and two dimensional, applications		Demonstration	4							

	4th week	4	Unit -3: Column chromatography 1. Column chromatography: Principle, classification, Experimental procedure, stationary and mobile		Power point Mid1	3					
			phases		Lecture	4					
	1st week	4	, development of the Chromatogram, applications.						1		
Jan	2nd week	4	2 HPLC: Basic principles, instrumentation –block diagram and applications	Types of HPLC	Lecture	3		Student seminar	•		
	3rd week	4	Unit 4: Spectrophotometry Principle, Instrumentation: Single beam and double beam spectrometer, Beer- Lambert's law- Derivation and deviations from Beer-Lambert's law,	Uses of spectrophotom etry.	Power point	3		Assignm ent	1		
	4th week	4	applications of Beer- Lambert's law-Quantitative determination of Fe+2, Mn+2and Pb+		Power point	4					
Feb	1st week	4	Unit -5: Atomic spectroscopy Types, atomizer, atomic absorption	Instrumentatio n	Lecture Mid2 ·	3					
	2nd week	4	and emission and applications		Demonstration	4					
	3rd week	4	Revision		Mind mapping	4					
	4th week	4	Revision		Question and answer method	2		Pre final	2		

SIGNATURE OF THE LECTURER

SIGNATURE OF THE HEAD OF THE DEPARTMENT

SIGNATURE OF THE PRINCIPAL